





<ul style="list-style-type: none"> <li>• 7) Introduction to the Chemistry Fair</li> <li>• 8) Test</li> <li>• 9) Feedback</li> </ul>	<ul style="list-style-type: none"> <li>• Apply your knowledge to discuss the chemistry associated with your model</li> <li>• Create a model to represent a type of bonding and its structure</li> <li>• Justify your choice of materials</li> <li>• Evaluate your knowledge.</li> <li>• Recognise areas of improvement and what went well.</li> <li>• Reflect on your knowledge.</li> <li>• Evaluate your knowledge.</li> <li>• Recognise areas of improvement and what went well.</li> <li>• Reflect on your knowledge.</li> </ul>		
<p><b><u>Chemistry: B) Chemical Reactions</u></b></p> <ul style="list-style-type: none"> <li>• 1) Signs of a chemical reaction</li> <li>• 2) Representing a Chemical Reaction</li> </ul>	<ul style="list-style-type: none"> <li>• Apply your knowledge to determine if a scenario is a chemical reaction or a physical change.</li> <li>• Summarise the different signs of a chemical reaction.</li> <li>• Conduct an experiment to demonstrate signs of a chemical reaction.</li> <li>• Write the molecular formula for simple substances.</li> <li>• Construct word equations that represent a chemical reaction.</li> </ul>	<ul style="list-style-type: none"> <li>• Working scientifically, application</li> <li>• Recall, application and working scientifically</li> <li>• Revise and reflect</li> <li>• Chemicals in the UAE</li> </ul>	<ul style="list-style-type: none"> <li>• Guided reading</li> </ul>

<ul style="list-style-type: none"> <li>• 3) Effect of Surface Area</li> </ul>	<ul style="list-style-type: none"> <li>• Construct chemical equations that represent a chemical reaction.</li> <li>• Use your knowledge to describe the conditions required for a reaction to take place.</li> <li>• Summarise how changing surface area effects the rate of a chemical reaction.</li> <li>• Conduct an experiment to demonstrate the effect of surface area on reaction rate.</li> </ul>		
<b>HALF TERM</b>			
<ul style="list-style-type: none"> <li>• 4) Effect of Temperature</li> <li>• 5) Reporting a Rate Investigation</li> </ul>	<ul style="list-style-type: none"> <li>• Apply your knowledge to describe the relationship between temperature and energy</li> <li>• Summarise how changing temperature effects the rate of a chemical reaction</li> <li>• Conduct an experiment to demonstrate the effect of temperature on reaction rate</li> <li>• Apply your knowledge to determine the variables for an investigation</li> <li>• Construct a table and/or graph to present your results</li> </ul>	<ul style="list-style-type: none"> <li>• Factories and catalysts in the UAE</li> </ul>	<ul style="list-style-type: none"> <li>• Guided Reading</li> </ul>

<ul style="list-style-type: none"> <li>• 6) Effect of a Catalyst</li> <li>• 7) Test</li> <li>• 8) Feedback</li> </ul>	<ul style="list-style-type: none"> <li>• Evaluate your results to write a valid conclusion supported by evidence</li> <li>• Use your knowledge to define the term catalyst</li> <li>• Summarise how adding a catalyst effects the rate of a chemical reaction</li> <li>• Compare the advantages of using a catalyst to changing other factors to increase the rate of a reaction</li> <li>• Evaluate your knowledge.</li> <li>• Recognise areas of improvement and what went well.</li> <li>• Reflect on your knowledge.</li> <li>• Evaluate your knowledge.</li> <li>• Recognise areas of improvement and what went well.</li> <li>• Reflect on your knowledge.</li> </ul>		
<p><b><u>Chemistry: C) Acids and Alkalis</u></b></p> <ul style="list-style-type: none"> <li>• 1) The pH Scale</li> <li>• 2) Identifying Acids and Alkalis</li> </ul>	<ul style="list-style-type: none"> <li>• Use your knowledge to describe the ions responsible for making a solution acid, alkali or neutral</li> <li>• Write the molecular formula for simple acids and alkalis</li> <li>• Calculate the RFM of simple acids and alkalis</li> </ul>	<ul style="list-style-type: none"> <li>• Recall, application and working scientifically</li> <li>• Water treatment in the UAE/ Comparing pH of water around the world</li> </ul>	<ul style="list-style-type: none"> <li>• Guided reading</li> </ul>

<ul style="list-style-type: none"> <li>• 3) Concentration</li>   <li>• 4) Neutralisation</li>   <li>• 5) Products of Neutralisation</li>   <li>• 6) Test</li> </ul>	<ul style="list-style-type: none"> <li>• Use your knowledge to describe the ions responsible for making a solution acid, alkali or neutral</li> <li>• Write the molecular formula for simple acids and alkalis</li> <li>• Calculate the RFM of simple acids and alkalis</li>   <li>• Use your knowledge to define the term concentration</li> <li>• Connect the concentration of an acid to its pH</li> <li>• Calculate the concentration of a solution using <math>n = c \times v</math></li>   <li>• Use your knowledge to state what is meant by neutralisation</li> <li>• Conduct an experiment which would demonstrate a neutralisation reaction</li> <li>• Predict if everyday substances are acid or alkali and give examples of real life neutralisation reactions</li>   <li>• Use your knowledge to name the salt produced in different neutralisation reactions</li> <li>• Write chemical equations for neutralisation reactions</li> <li>• Evaluate separation techniques to determine the best method for collect salts from water</li>   <li>• Evaluate your knowledge.</li> </ul>		
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<ul style="list-style-type: none"> <li>7) Feedback</li> </ul>	<ul style="list-style-type: none"> <li>Recognise areas of improvement and what went well.</li> <li>Reflect on your knowledge.</li> <li>Evaluate your knowledge.</li> <li>Recognise areas of improvement and what went well.</li> <li>Reflect on your knowledge.</li> </ul>		
<p><b><u>Chemistry: D) Real world Chemistry</u></b></p> <ul style="list-style-type: none"> <li>1) Chemistry for Fuels</li> <li>2 &amp; 3) Renewable Energy</li> <li>4) Chemistry &amp; Farming</li> <li>5) Chemistry &amp; Plastics</li> </ul>	<ul style="list-style-type: none"> <li>Apply your knowledge to discuss the process of oil forming</li> <li>Summarise the process of combustions</li> <li>Evaluate the use of oil as a fuel</li> <li>Use your knowledge to describe the formation of bioethanol</li> <li>Construct the chemical equation for the formation of biofuel</li> <li>Evaluate the use of bioethanol and its impact on the environment and society</li> <li>Use your knowledge to discuss the structure and bonding of ammonia</li> <li>Summarise the process of making ammonia for use in fertilisers</li> <li>Justify why fertilisers are an important in the UAE and evaluate their use</li> <li>Use your knowledge to define the term monomer and polymer</li> </ul>	<ul style="list-style-type: none"> <li>Working scientifically, apply, evaluate</li> <li>Renewable energy resources in the UAE</li> <li>Fossil fuels (oil) in the UAE</li> <li>Plastic production in the UAE</li> </ul>	<ul style="list-style-type: none"> <li>Guided reading</li> </ul>

<ul style="list-style-type: none"> <li>• 6) Analytical Chemistry</li>   <li>• 7) Test</li>   <li>• 8) Feedback</li> </ul>	<ul style="list-style-type: none"> <li>• Summarise the process of polymerisation and represent using an equation</li> <li>• Debate the use of biodegradable and non-biodegradable plastics</li>   <li>• Use your knowledge to explain how paper chromatography works</li> <li>• Conduct paper chromatography to separate coloured substances</li> <li>• Interpret results of paper chromatography to compare and contrast different inks</li>   <li>• Evaluate your knowledge.</li> <li>• Recognise areas of improvement and what went well.</li> <li>• Reflect on your knowledge.</li>   <li>• Evaluate your knowledge.</li> <li>• Recognise areas of improvement and what went well.</li> <li>• Reflect on your knowledge.</li> </ul>		
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Winter Break